

10 → Start and explain Fourier's heat flow equation.

⇒ Fourier's heat flow → If ~~any~~ a solid of volume V be enclosed by a surface S , the temperature of which is not the same at all ~~pp~~ points. Heat would thus flow from places at higher temperature to those at lower by conduction. The amount of heat flowing out of a small surface ds in time Δt is

$$dQ = -[k \vec{\nabla} \theta \cdot \hat{n} ds] \Delta t$$

Where k is thermal conductivity, θ the temperature at (x, y, z, t) and \hat{n} the outward unit ~~is~~ S (is normal) ^x

∴ Total amount of heat flowing out of the whole surface S in time Δt is

$$Q = -\Delta t \iint_S k \vec{\nabla} \theta \cdot \hat{n} ds$$

$$= -\Delta t \iiint_V \vec{\nabla} \cdot (k \vec{\nabla} \theta) dv \quad \text{--- (i)}$$

By the divergence theorem.

If c be the specific heat of the material, ρ its density and $\partial\theta/\partial t$ the rate of increase of temperature within the volume V of the body, the amount of heat flowing out in time Δt is (out of V)

$$Q = -\Delta t \iiint_V \rho c \frac{\partial \theta}{\partial t} dv \quad \text{--- (ii)}$$

from (i) and (ii), therefore

$$\iiint_V \left\{ \rho c \frac{\partial \theta}{\partial t} - \vec{\nabla} \cdot (k \vec{\nabla} \theta) \right\} dv = 0$$

Since dv is arbitrary, $\rho c \frac{\partial \theta}{\partial t} = \vec{\nabla} \cdot (k \vec{\nabla} \theta)$

$$\text{or, } \boxed{\frac{\partial \theta}{\partial t} = \frac{k}{\rho c} \nabla^2 \theta = h \nabla^2 \theta} \quad \text{--- (iii)}$$

When $h = k/\rho c$ is the thermal diffusivity of material. eqn (iii) is the three-dimensional Fourier's heat flow eqⁿ